



Fig. 1. Monoclinic cell parameters (\AA) of RE NbO_4 and ionic radii of the corresponding eight-coordinated trivalent rare earth (after Shannon & Prewitt, 1969, 1970).

in oxides. The contraction of the rare earths' radii is essentially linear, except for positive deviations at La and Lu. This deviation from linearity is reflected in the cell parameters of the RE NbO_4 compounds.

The help of William Holt who prepared several of the compounds, and Clarence Gooden who assisted with the data refinement, is acknowledged.

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Acta Cryst. (1971) **B27**, 2286

Note on the space group of potassium hydrogeniodate(V), $\text{KIO}_3 \cdot \text{HIO}_3$. A correction. By G. KEMPER and AAFJE VOS, *Laboratorium voor Structuurchemie, Rijksuniversiteit Groningen, Zernikelaan, Paddepoel, Groningen, The Netherlands*

(Received 24 July 1971)

Contrary to the conclusion given in an earlier note, we have adopted the space group $P2_1/c$ for $\text{KIO}_3 \cdot \text{HIO}_3$, as the observed piezoelectric effect appeared to be caused by a thin layer of KIO_3 present on the surface of the crystal of $\text{KIO}_3 \cdot \text{HIO}_3$ used for the piezoelectricity measurement.

The difficulties concerning the space group of $\text{KIO}_3 \cdot \text{HIO}_3$ reported in an earlier note (Kemper & Vos, 1970) have now been solved. It appeared that the piezoelectric effect which had been measured for a large crystal of $\text{KIO}_3 \cdot \text{HIO}_3$ having a volume of approximately 0.25 mm^3 , is caused by a thin layer of KIO_3 deposited on the surface of the crystal. The presence of the KIO_3 layer was found by a careful study of a zero level Weissenberg photograph about the b axis of the large crystal, taken with Ni-filtered Cu radiation. This

Table 2. X-ray powder pattern of TmNbO_4

d_{obs}	d_{calc}	I/I_o	hkl
5.44	5.44	2	020
4.718	4.720	2	110
4.562	4.563	9	011
3.106	3.105	100	121
2.981	2.980	10	130
2.939	2.939	90	031, 121
2.718	2.717	20	040
2.620	2.621	20	200
2.514	2.513	25	002
2.285	2.286	2	112, 022
2.206	2.205	7	141, 211
2.145	2.144	7	141
2.005	2.006	2	150, 231
1.996	1.996	9	051
1.966	1.966	3	132
1.891	1.891	3	202
1.8865	1.8867	40	240
1.8455	1.8458	30	042
1.7457	1.7460	12	202
1.6564	1.6562	2	013
1.6344	1.6343	16	103, 161
1.6161	1.6163	14	321
1.6083	1.6086	24	161
1.5737	1.5735	2	330
1.5654	1.5657	13	123
1.5524	1.5526	9	242
1.5444	1.5441	13	152, 321
1.4999	1.5000	11	123

photograph showed 13 weak streaks which were identified as KIO_3 reflexions by their good fit to the pattern of a zero-layer Weissenberg photograph of the room temperature modification of KIO_3 at present being studied by F. van Bolhuis in our laboratory. We have therefore adopted $P2_1/c$ as the correct space group for $\text{KIO}_3 \cdot \text{HIO}_3$ in agreement with the results of the X-ray refinement reported in the earlier note (Kemper & Vos, 1970). In the mean time a neutron powder-diffraction study of $\text{KIO}_3 \cdot \text{HIO}_3$ and of $\text{KIO}_3 \cdot \text{DIO}_3$

carried out by Dr H. M. Rietveld at the Reactor Centrum Nederland, Petten (N.H.), has shown that the hydrogen (deuterium) atoms also obey the symmetry P_{21}/c .

The crystal structure of $\text{KIO}_3 \cdot \text{HIO}_3$ has recently also been determined by Chan & Einstein (1971). In view of the discussion of the piezoelectric effect given above, it is reasonable to assume that the strong second harmonic effect reported for $\text{KIO}_3 \cdot \text{HIO}_3$ by these authors is due to the presence of KIO_3 (or HIO_3) on the surface of the crystals.

A full account of our work and a comparison with the structure of $\text{KIO}_3 \cdot \text{HIO}_3$ described in space group P_{21}/c by Chan & Einstein will be published in the *Canadian Journal of Chemistry*.

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Acta Cryst. (1971). **B27**, 2287

A redetermination of the crystal structure of tetramethyldiphosphine disulphide. By J. D. LEE and G. W. GOODACRE, Department of Chemistry, University of Technology, Loughborough, Leicestershire, England

(Received 31 May 1971)

The correct structure factor table to an earlier paper [*Acta Cryst.* (1971) **B27**, 302] is given.

In an earlier paper of the above title (Lee & Goodacre, 1971) the two parts of the structure factor table printed (Table 4, page 305) are identical. The whole Table should be replaced by that given here.

Table 4. Observed and calculated structure factors

h	k	l	F_O	F_C	h	k	l	F_O	F_C	h	k	l	F_O	F_C	h	k	l	F_O	F_C				
-2	0	0	2.0	1.9	0	1	7.9	62.7	59.0	1	10.1	17.8	-18	8	1	29.1	-27.9	16	6	10.5	11.6		
-4	0	0	1.9	1.8	0	1	10.1	34.2	32.0	1	10.1	11.2	-1	7	2	41.1	-40.7	16	6	10.5	11.6		
-6	0	0	1.7	1.6	0	1	10.1	34.2	32.0	1	10.1	11.2	-1	7	2	41.1	-40.7	16	6	10.5	11.6		
-12	0	0	129.7	129.7	1	1	56.0	-26.1	51.1	1	56.1	-31.1	-5	9	1	19.1	16.0	7	7	5.8	6.6		
-14	0	0	8.8	8.1	1	1	2.7	21.1	1	2	22.7	19.8	-7	9	1	36.3	-34.1	17	7	2.2	2.0		
-18	0	0	33.3	33.3	2	2	26.1	2.1	1	1	1.7	1.1	-1	9	1	12.7	13.0	17	7	1.7	1.0		
-20	0	0	0.5	0.5	2	2	1.1	1.1	1	1	1.7	1.1	-1	9	1	12.7	13.0	17	7	1.7	1.0		
-1	-1	0	2.9	2.9	1	1	15.0	18.1	-1	1	15.0	18.1	-1	9	1	6.5	6.7	15	7	2.2	2.7		
-1	-1	0	17.3	17.6	6	2	3.3	17.3	-2.1	1	15.0	18.1	-1	9	1	6.5	6.7	15	7	2.2	2.7		
-2	-1	0	1.1	1.1	1	1	15.0	17.3	-2.1	1	15.0	18.1	-1	9	1	6.5	6.7	15	7	2.2	2.7		
-3	-1	0	111.4	127.0	6	2	2.1	8.1	-1	1	12.4	12.1	-1	9	1	5.9	5.8	15	7	1.7	1.0		
-4	-1	0	1.1	1.1	1	1	15.0	17.3	-2.1	1	15.0	18.1	-1	9	1	6.5	6.7	15	7	2.2	2.7		
-13	-1	0	4.4	11.9	8	2	1	36.1	12.3	-1	10	10	15.5	2	8	10.9	10.9	14	7	5.6	6.8		
-15	-1	0	32.6	32.6	7	2	37.0	12.2	-1	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8		
-17	-1	0	7.7	8.1	12	1	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8					
-19	-1	0	9.3	9.9	14	2	1	15.6	15.3	-2	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8	
-20	-2	0	3.8	4.8	12	1	1	15.6	15.3	-2	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8	
-2	-2	0	4.6	2.8	2	2	12.0	17.5	-1	2	33.7	21.7	-7	11	2	12.0	21.7	14	7	5.6	6.8		
-4	-2	0	7.0	8.9	2	2	1.1	17.6	-1	2	24.6	23.1	-1	8	2	9.3	9.3	14	7	5.6	6.8		
-4	-2	0	14.4	14.4	1	1	15.6	15.3	-2	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8		
-12	-2	0	147.2	147.2	8	18	0.5	1	15.6	15.3	-2	1	15.6	15.3	-2	8	2	17.1	19.2	14	7	5.6	6.8
-18	-2	0	17.9	18.9	5	3	50.6	65.9	25.0	9	1	5.2	2.4	-2	12	1	5.9	5.8	15	7	2.2	2.7	
-20	-2	0	14.4	14.7	7	2	49.2	46.7	2.1	1	5.6	5.1	-1	10	1	5.6	6.8	15	7	2.2	2.7		
-9	-3	0	178.5	183.1	1	1	5.6	42.6	-1	1	5.6	42.6	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-11	-1	0	1.7	1.7	1	1	6.6	6.5	-2	1	6.6	6.5	-2	1	1	12.0	12.0	14	7	5.6	6.8		
-13	-1	0	1.7	1.7	1	1	12.7	14.3	-1	1	12.7	14.3	-1	1	1	17.2	16.6	14	7	5.6	6.8		
-15	-1	0	1.7	1.7	1	1	12.7	14.3	-1	1	12.7	14.3	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-19	-1	0	8.9	7.5	19	2	17.0	18.6	-1	1	17.0	18.6	-1	1	1	2.2	2.7	14	7	5.6	6.8		
-21	-2	0	1.7	1.7	23	2	17.0	18.6	-1	1	17.0	18.6	-1	1	1	2.2	2.7	14	7	5.6	6.8		
-2	-2	0	1.7	1.7	23	2	17.0	18.6	-1	1	17.0	18.6	-1	1	1	2.2	2.7	14	7	5.6	6.8		
-2	-2	0	4.8	5.5	2	4	2.1	2.1	-2	1	20.9	21.8	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-4	-2	0	5.0	5.0	1	1	2.1	2.1	-2	1	20.9	21.8	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-6	-2	0	5.0	5.0	3	3	1	1	1	1	21.0	21.9	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-8	-2	0	5.0	5.0	3	3	1	1	1	1	21.0	21.9	-1	1	1	12.0	12.0	14	7	5.6	6.8		
-12	-2	0	131.7	122.7	10	4	33.4	37.8	-8	2	2.2	1.7	-2	16.0	-1	4	69.1	79.5	14	7	5.6	6.8	
-14	-2	0	7.4	9.8	12	4	1.1	0.2	-1	1	29.4	21.9	-1	2	10.0	9.8	14	7	5.6	6.8			
-5	-5	0	5.6	5.6	16	1	16.7	16.6	-2	1	20.6	19.8	-2	1	16.0	15.5	14	7	5.6	6.8			
-5	-5	0	12.8	11.8	16	1	16.7	16.6	-2	1	20.6	19.8	-2	1	16.0	15.5	14	7	5.6	6.8			
-19	-1	0	8.9	7.5	19	2	17.0	18.6	-1	1	17.0	18.6	-1	1	2.2	2.7	14	7	5.6	6.8			
-21	-2	0	1.7	1.7	23	2	17.0	18.6	-1	1	17.0	18.6	-1	1	2.2	2.7	14	7	5.6	6.8			
-2	-2	0	1.7	1.7	23	2	17.0	18.6	-1	1	17.0	18.6	-1	1	2.2	2.7	14	7	5.6	6.8			
-2	-2	0	4.8	4.8	1	1	17.0	18.6	-1	1	17.0	18.6	-1	1	2.2	2.7	14	7	5.6	6.8			
-15	-5	0	12.0	11.7	1	1	5.1	5.1	-5	1	14.3	14.3	-18	1	12.0	11.7	14	7	5.6	6.8			
-15	-5	0	14.3	14.6	3	5	1	1	5.1	5.1	-5	1	14.3	14.3	-18	1	12.0	11.7	14	7	5.6	6.8	
-17	-5	0	110.5	111.1	-7	3	1	6.1	7.4	-1	1	6.1	7.4	-1	2	12.0	11.7	14	7	5.6	6.8		
-9	-5	0	160.7	173.0	1	1	5.9	10.0	-1	1	14.5	14.5	-22	1	12.0	11.7	14	7	5.6	6.8			
-2	-6	4	4.1	7.5	7	5	2	36.2	39.1	-1	1	14.1	13.1	-1	1	12.0	11.7	14	7	5.6	6.8		
-6	-6	0	1.7	7.5	9	5	1	1	14.1	13.1	-1	1	14.1	13.1	-1	1	12.0	11.7	14	7	5.6	6.8	
-12	-8	0	90.7	88.1	15	3	1	13.9	31.4	-1	1	9.4	10.4	-1	2	12.0	11.7	14	7	5.6	6.8		
-14	-6	0	7.6	16.6	15	3	1	13.9	31.4	-1	1	24.2	42.0	-5	1	12.0	11.7	14	7	5.6	6.8		
-16	-6	0	21.6	21.5	15	5	1	13.9	31.4	-1	1	24.2	42.0	-5	1	12.0	11.7	14	7	5.6	6.8		
-18	-8	0	21.5	21.6	16	6	1	13.9	31.4	-1	1	24.2	42.0	-5	1	12.0	11.7	14	7	5.6	6.8		
-2	-9	0	4.4	4.7	1	1	13.9	31.4	-1	1	24.2	42.0	-5	1	12.0	11.7	14	7	5.6	6.8			
-9	-9	0	51.6	51.3	3	7	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	11	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17	-9	0	18.3	18.2	12	-1	1	8.3	7.5	-1	1	10.6	11.1	-1	2	20.2	20.7	14	7	5.6	6.8		
-17																							